What is claimed is:

1. A method of lowering the melting temperature of a glass composition including CaO and MgO while substantially maintaining the bending and annealing temperatures, comprising the steps of:

increasing the CaO by a selected weight percent; and decreasing the MgO by substantially the same weight percent.

- 2. The method according to claim 1, including increasing the CaO to greater than 9 weight percent.
- 3. The method according to claim 1, including increasing the CaO to 9.1 to 12 weight percent.
- 4. The method according to claim 1, including increasing the CaO to greater than or equal to 10 weight percent.
- 5. The method according to claim 1, including decreasing the MgO to less than 3 weight percent.
- 6. The method according to claim 1, including maintaining a total amount of CaO + MgO greater than 12 weight percent.
- 7. The method according to claim 1, including maintaining a total amount of CaO + MgO greater than 12.5 weight percent.
- 8. The method according to claim 1, including maintaining a total amount of CaO + MgO in the range of 12.5 to 13 weight percent.

- 9. The method according to claim 1, including increasing the CaO to provide a melting temperature in the range of about 2500°F to about 2590°F (1370°C to 1421°C), a bending temperature in the range of about 1300°F to 1400°F (704°C to 759°C), and an annealing temperature in the range of about 1010°F to 1050°F (543°C to 565°C).
- 10. A method of adjusting a glass composition to lower the melting and forming temperatures while substantially maintaining the bending and annealing temperatures, comprising the steps of:

providing a glass composition having CaO and MgO; increasing the CaO a selected amount; and decreasing the MgO by substantially the same selected amount while substantially maintaining a total amount of CaO + MgO.

11. A method of lowering the melting and forming temperatures of a glass composition while substantially maintaining the softening and annealing temperatures of the glass, comprising:

replacing at least a portion of at least one of CaO or MgO in the composition with a metal oxide whose metal has a lower field strength than at least on e of Ca^{++} or Mg^{++} .

12. The method according to claim 11, including replacing at least a portion of at least one of the CaO or MgO with at least one metal oxide whose metal is selected from Ba or Sr.

A glass composition, comprising: 13. 70 to 75 weight percent SiO_2 12 to 15 weight percent Na₂O 0 to 5 weight percent K_2O > 9 weight percent CaO MgO < 4 weight percent 0 to 2 weight percent Al₂O₃ SO_3 f to 1 weight percent 0 to 2 weight percent Fe_2O_3 wherein: $SiO_2 + Al_2O_3$ ≥ 70 weight percent $Na_2O/+ K_2O$ 10 to 15 weight percent Ca/O + MgO 12 to 15 weight percent aO/MgO 2 to 5/

14. The composition according to claim 13, wherein Cap is in the range of greater than 9 to 12 weight percent.

- 15. The composition according to claim 13, wherein CaO is in the range of 9.1 to 11 weight percent.
- 16. The composition according to claim 13, wherein MgO is in the range of 2 to less than 4 weight percent.
- 17. The composition according to claim 13, wherein CaO + MgO is in the range of 12 to 13.5 weight percent.
- 18. The composition according to claim 13, wherein CaO + MgO is in the range of 12.5 to 13 weight percent.
- 19. The composition according to claim 13, wherein the glass composition has a log 2 viscosity in the range of about 2570°F to about 2590°F (1410°C to 1421°C) and a log 4

viscosity in the range of about 1850° F to about 1894° F (1010°C to 1034° C).

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- 20. The composition according to claim 13, wherein the glass composition has a log 7.6 viscosity in the range of about 1300°F to about 1350°F (704°C to 732°C) and a log 13 viscosity in the range of about 1016°F to about 1020°F (547°C to 549°C).
- 21. The composition according to claim 19, wherein the glass composition has a log 7.6 viscosity in the range of about 1300°F to about 1350°F (704°C to 732°C) and a log 13 viscosity in the range of about 1016°F to about 1020°F (547°C to 549°C).

A flat glass product made by the process of

claim 1.

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